

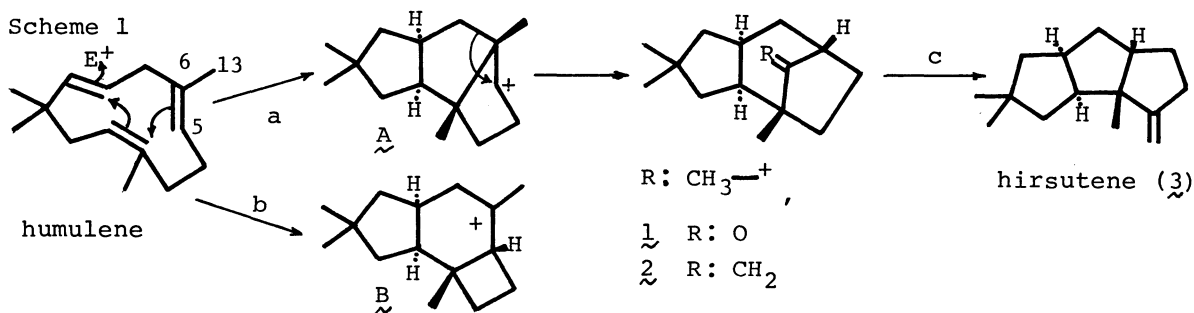
## A CONVERSION OF HUMULENE TO HIRSUTENE

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Humulene has been transformed into 1,4,4-trimethyltricyclo-[6.2.1.0<sup>2,6</sup>]undecan-11-one, a key intermediate for the synthesis of hirsutene, through a process which simulates the biosynthetic pathway assumed for the latter sesquiterpene.

In the previous paper<sup>1)</sup> we reported chemical synthesis of tricyclic compound 2 and its rearrangement to hirsutene 3 through a biogenetic-like path (Scheme 1, c)<sup>2)</sup> Compound 2 in turn was prepared from its norketone 1. In another paper we showed<sup>3)</sup> that humulene can be converted to protoilludyl cation B under suitable conditions, but not to secondary cation A, presumably because of the presence of the 13-methyl group. It was therefore anticipated that a properly substituted 13-norhumulene would give an A type cation rather than a B type cation. We should like to describe herein transformation of humulene into the norketone 1 through 13-nor-cation of A. The transformation means the success of the first, even if not direct, synthesis of tricyclic hirsutene from its monocyclic biogenetic precursor.

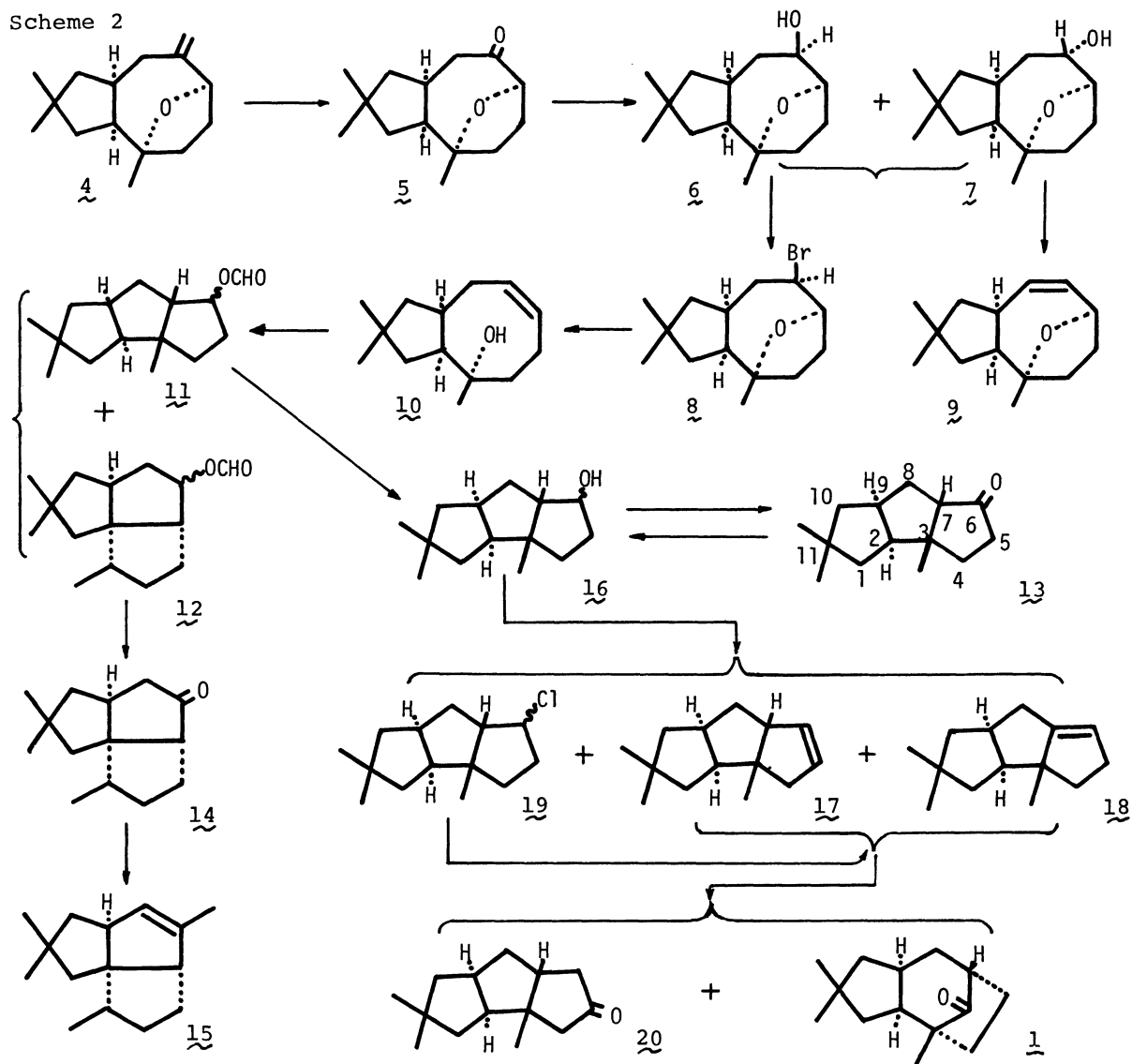


A 3,6-secoprotoilludene derivative 4, which was previously derived from humulene by us,<sup>3)</sup> was first converted to 5<sup>4)</sup> ( $OsO_4-NaIO_4/tBuOH-H_2O/rt/2\text{ hr}$ , 95%). The ketone was then reduced to a 1:1 mixture of epimeric alcohols, 6 and 7 (DIBAH/THF/-78°/10 min, almost quantitative) which was separated by passing through a silica gel column [6<sup>4)</sup>: mp 108-110 °C<sup>5)</sup>;  $ir(CCl_4, 10^{-3}\text{ mol})$  3610  $cm^{-1}$ , 7<sup>4)</sup>: oil;  $ir(CCl_4, 10^{-3}\text{ mol})$  3515  $cm^{-1}$ ]. Configuration of the hydroxyl group of each

isomer was determined by the presence of an intramolecular H-bond in the ir spectrum of 7. Treatment of 6 with  $\text{CBr}_4/\phi_3\text{P}$  gave  $\beta$ -bromide 8<sup>4)</sup> ( $\phi\text{H}/\text{reflux}/5$  min, 95%) (mp 91-2°). The  $\beta$ -configuration of the bromine atom was deduced from resemblance of the nmr coupling constants of the proton at C-6 (dt, 11, 4 Hz) to those of the C-6 proton of 6 (dt 10, 5 Hz). On the other hand,  $\alpha$ -ol 7 gave exclusively olefin 9<sup>4)</sup> on the similar treatment (95%). Cleavage of the ether linkage of 8 ( $\text{Zn-EtOH}/\text{reflux}/12$  hr) afforded 10<sup>4)</sup> (90%). The nmr spectrum of 10 in the presence of  $\text{Eu}^{3+}$  [ $\text{Eu}(\text{fod})_3/\text{10}=0.51$ ,  $\text{CCl}_4$ ] showed two double triplets at  $\delta$  7.75 ( $J=6, 10$  Hz, 6-H) and at 7.25 ( $J=7, 10$  Hz, 7-H). Therefore the configuration of the double bond was assigned to be cis on the basis of the coupling constant 10 Hz. The cyclooctenol 10 was heated to reflux in  $\text{HCO}_2\text{H-Ac}_2\text{O}$  for 30 min under argon to give rise to a mixture of formates, 11 and 12 (80%, 3:1(glc)). The formates were treated as such with  $\text{LiAlH}_4$  (THF/reflux/10 min, nearly quantitative) and the product was next oxidized with Jones reagent (0°/15 min, 95%) to yield a mixture of ketones (13<sup>4)</sup> and 14<sup>4)</sup>, 3:1(glc), which was separated into two isomers by silica gel chromatography (13:ir 1745  $\text{cm}^{-1}$ ; nmr  $\delta$  0.92(3H, s), 1.06(6H, s), 14: ir 1740  $\text{cm}^{-1}$ , nmr  $\delta$  0.99(3H, d, 7 Hz), 1.02, 1.10(each 3H, s)].

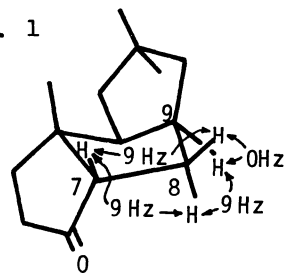
Structure of 13 was revealed by nmr decoupling experiments in the presence of a shift reagent [ $\text{Eu}(\text{fod})_3/\text{13}=0.3$ ,  $\text{CCl}_4$ ,  $\delta$  7.01( $S=11.87$ , 1H, d,  $J_{4\alpha-5\alpha}=8$  Hz, 5 $\alpha$ H), 6.97( $S=11.17$ , 1H, dd,  $J_{4\beta-5\beta}=8$  Hz,  $J_{4\alpha-5\beta}=4$  Hz, 5 $\beta$ H), 6.55( $S=11.17$ , 1H, t,  $J_{7\beta-8\alpha}=J_{7\beta-8\beta}=9$  Hz, 7 $\beta$ H), 4.85( $S=7.85$ , 1H, dt,  $J_{7\beta-8\alpha}=J_{8\alpha-9\alpha}=9$  Hz,  $J_{8\beta-8\alpha}=13$  Hz, 8 $\alpha$ H), 3.95( $S=2.61$ , 1H, m, 9 $\alpha$ H), 3.64( $S=5.48$ , 1H, dd,  $J_{8\alpha-8\beta}=13$  Hz,  $J_{7\beta-8\beta}=9$  Hz, 8 $\beta$ H), 3.20( $S=3.91$ , 1H, ddd,  $J_{4\alpha-4\beta}=12$  Hz,  $J_{4\beta-5\alpha}=5$  Hz,  $J_{4\beta-5\beta}=8$  Hz, 4 $\beta$ H), 2.28( $S=3.22$ , 3H, s, 12Me), 2.25( $S=1.00$ , 1H, dd,  $J_{9\alpha-10\alpha}=7$  Hz,  $J_{10\alpha-10\beta}=13$  Hz, 10 $\alpha$ H), 2.05( $S=1.20$ , 2H, d,  $J_{1\alpha,\beta-2\alpha}=9$  Hz, 1 $\alpha,\beta$ H), 1.82( $S=1.83$ , 1H, dd,  $J_{9\alpha-10\beta}=8$  Hz,  $J_{10\alpha-10\beta}=13$  Hz, 10 $\beta$ H). Moreover, the large  $S$  value of 9 $\alpha$ H in comparison with that of 8 $\beta$ H and the observed  $J$  values ( $J_{7\beta-8\alpha}=J_{7\beta-8\beta}=J_{8\alpha-9\alpha}=9$  Hz,  $J_{8\beta-9\alpha}=0$  Hz) indicated a cis-anti-cis stereochemistry as depicted in Fig. 1 for this ketone. On the other hand, structure of 14 was determined by leading it to the known compound, pentalenene, 15<sup>7)</sup> [(1)  $\text{MeMgI}/\text{ether}/\text{reflux}/2$  hr, 90%, (2)  $\text{MsCl}/\text{py}/\text{rt}/12$  hr, 90%]. Reduction of 13 ( $\text{NaBH}_4/\text{ether}/0^\circ/10$  min) furnished a pair of epimeric alcohols 16<sup>4)</sup> (quantitative) which in turn was treated with  $\text{MsCl-py}$  (100°/30 min) to give epimeric chlorides (34%) and two olefins 17 and 18 (42%, 9:1, glc). The chlorides were separated from the olefins by column chromatography ( $\text{SiO}_2$ ) (19: nmr

Scheme 2



$\delta$  0.92(3H, s), 1.08(6H, s), 3.86 and 4.28(1H, each m), 40%) and were further dehalogenated ( $t$ AmONa/DMSO/80°/8 hr) to afford a similar mixture of the olefins, **17** and **18** (9:1(glc)). Since the olefins could not be separated by the usual methods, they were treated as such with  $Tl(ClO_4)_3$  ( $t$ BuOH-H<sub>2</sub>O(1:1)/rt/3 hr) to give two ketones (53%, **20**/**1** = 5/1, silica gel column, **20**: ir 1745 cm<sup>-1</sup>, nmr  $\delta$  0.91, 1.02, 1.07(each 3H, s), **1**: ir 1750 cm<sup>-1</sup>; nmr  $\delta$  0.90, 0.98, 1.10(each 3H, s)). Stereostructure of **20** was deduced from the decoupling experiments of the nmr spectrum in the presence

Fig. 1



of  $\text{Eu}^{3+}$ : ( $\text{Eu}(\text{fod})_3/\text{20} \approx 0.35$ , ( $\text{CCl}_4$ )  $\delta$  8.22 (S=15.40, 1H, dd,  $J_{6\alpha-6\beta}=19$  Hz,  $J_{6\beta-7\beta}=6$  Hz), 6 $\beta$ H), 7.82 (S=15.00, 1H, dd,  $J_{6\alpha-6\beta}=19$  Hz,  $J_{6\alpha-7\beta}=2.5$  Hz, 6 $\alpha$ H), 7.80 and 7.68 (S=15.20 and 14.60, 2H, ABq,  $J_{4\alpha-4\beta}=19$  Hz, 4 $\alpha$  and 4 $\beta$ H), 4.40 (S=4.52, 1H, m, 7 $\beta$ H), 4.10 (S=3.12, 2H, m, 9 $\alpha$ , 8 $\alpha$ H), 4.00 (S=5.04, 1H, dt,  $J_{1\alpha-2\alpha}=14$  Hz,  $J_{1\beta-2\alpha}=J_{2\alpha-9\alpha}=9$  Hz, 2 $\alpha$ H), 3.20 (S=2.40, 1H, dd,  $J_{8\alpha-8\beta}=12$  Hz,  $J_{7\beta-8\beta}=8$  Hz, 8 $\beta$ H), 2.92 (S=4.20, 3H, s, 12Me), 1.72 (S=0.92, 3H, s, 14Me), 1.60 (S=1.08, 3H, s, 14Me). Ketone 1 was identified with an authentic sample of 13-norprotophirsutan-7-one, which had already been transformed to hirsutene<sup>1)</sup>.

#### References

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3. S. Misumi, T. Ohtsuka, Y. Ohfuné, K. Sugita, H. Shirahama, and T. Matsumoto, *Tetrahedron Lett.*, 1979, 31; S. Misumi, T. Ohtsuka, H. Hashimoto, Y. Ohfuné, H. Shirahama, and T. Matsumoto, *Tetrahedron Lett.*, 1979, 35.
4. Satisfactory spectral data were obtained for this compound.
  - 5: ir 1710  $\text{cm}^{-1}$ ; nmr  $\delta$  1.01, 1.08, 1.20 (each 3H, s), 4.15 (1H, d, 8 Hz).
  - 6: nmr  $\delta$  0.97, 1.04, 1.16 (each 3H, s), 3.62 (1H, dt, 10, 5 Hz), 4.08 (1H, m).
  - 7: nmr  $\delta$  0.99, 1.04, 1.19 (each 3H, s), 3.40 (1H, dt, 5, 2.5 Hz), 4.20 (1H, m).
  - 8: ir 1070, 1055  $\text{cm}^{-1}$ ; nmr  $\delta$  0.95, 1.02, 1.13 (each 3H, s), 4.00 (1H, dt, 11, 4 Hz), 4.35 (1H, m).
  - 9: ir 1650, 1090, 1065  $\text{cm}^{-1}$ ; nmr  $\delta$  0.99, 1.05, 1.11 (each 3H, s), 4.50 (1H, m), 5.25 (1H, dd, 2, 12 Hz), 5.35 (1H, d, 12 Hz).
  - 10: ir 3400, 1100  $\text{cm}^{-1}$ ; nmr  $\delta$  1.01, 1.11, 1.13 (each 3H, s), 5.80 (2H, m).
  - 13: ir 1743  $\text{cm}^{-1}$ ; nmr  $\delta$  0.90 (3H, s), 1.04 (6H, s).
  - 14: ir 1740  $\text{cm}^{-1}$ ; nmr  $\delta$  0.97 (3H, d,  $J=6.5$  Hz), 1.02, 1.10 (each 3H, s).
  - 16:(a) ir 3100, 1070  $\text{cm}^{-1}$ ; nmr  $\delta$  0.92, 0.97, 1.01 (each 3H, s), 4.24 (1H, m).
  - (b) nmr  $\delta$  0.92, 1.07, 1.12 (each 3H, s), 3.83 (1H, m).  $\frac{16a}{16b}=5/4$ .
5. Mps were noted for all crystalline compounds.
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